

The role of certain ions in the functioning of inorganic pyrophosphatase, one of the key enzymes of the phosphorus metabolism, has been little studied, particularly for the vegetable kingdom, including the cotton plant. The aim of the present work was to study the influence of certain ions on the hydrolytic activity of cotton pyrophosphatase [1].

Influence of  $\text{Co}^{2+}$  and  $\text{Mn}^{2+}$ . In an investigation of the action of  $\text{Co}^{2+}$  and  $\text{Mn}^{2+}$  on the hydrolysis of  $\text{PP}_i$  by cotton pyrophosphatase, activation of the enzyme was detected, as in the case of the ions  $\text{Mg}^{2+}$  [2] and  $\text{Zn}^{2+}$  [3]. It must be mentioned that the  $\text{Co}^{2+}$  and  $\text{Mn}^{2+}$  ions, unlike  $\text{Mg}^{2+}$  and  $\text{Zn}^{2+}$ , stimulate inorganic pyrophosphatase at comparatively high concentrations of  $\text{PP}_i$ , (8 mM) in the incubation medium, the optimum concentrations for  $\text{CoCl}_2$  and  $\text{MnCl}_2$  being 0.5 mM. A further rise in the concentration of  $\text{Co}^{2+}$  and  $\text{Mn}^{2+}$  in the incubation medium appreciably inhibited the pyrophosphatase reaction. However, inhibition by an excess of  $\text{Co}^{2+}$  was weaker than by an excess of  $\text{Mn}^{2+}$  in the medium. This phenomenon may apparently be connected with the complication of the formation of polymeric complexes in the  $\text{Mn}^{2+}$ - $\text{PP}_i$  system, as has been established for yeast pyrophosphatase [4].

Influence of  $\text{Ca}^{2+}$ ,  $\text{Cu}^{2+}$ , and  $\text{Sr}^{2+}$ . With respect to the enzyme under investigation,  $\text{Ca}^{2+}$  ions occupy a neutral position or they inhibit the hydrolytic capacity of cotton pyrophosphatase. Investigations of the action of  $\text{Cu}^{2+}$  and  $\text{Sr}^{2+}$  on the activity of the enzyme have shown that these ions are inhibitors of the pyrophosphatase reaction, while the inhibiting capacity of  $\text{Cu}^{2+}$  is almost an order of magnitude greater than that of  $\text{Sr}^{2+}$ . The inhibition constant ( $K_i$ ) of the magnesium-activated reaction of the pyrophosphatase from cotton-plant shoots was determined from the dependence of the rate on the concentration of copper: it proved to be 90  $\mu\text{M}$ .

Influence of  $\text{F}^-$ . The results of a study of the influence of fluoride ions on pyrophosphatase activity has shown that fluoride ions also inhibit the enzymatic activity, lowering  $K_i$  and  $V$ . It must be mentioned that this inhibition of cotton pyrophosphatase by  $\text{F}^-$  belongs to the noncompetitive type of inhibition. Hence, it may be assumed that fluorine is capable of adding only to the enzyme-substrate complex and not to the free pyrophosphatase enzyme, which is more characteristic for the noncompetitive inhibition of enzymatic reactions. As a result of this, the inhibitor strongly stabilizes the intermediate compound of the protein with a  $\text{PP}_i$  molecule, as has been shown in the case of yeast pyrophosphatase [4]. The value of  $K_i$  for  $\text{F}^-$  ions amounts to 420  $\mu\text{M}$ .

Summarizing what has been said above, it may be observed that, in the first place,  $\text{Mg}^{2+}$ ,  $\text{Zn}^{2+}$ ,  $\text{Co}^{2+}$ , and  $\text{Mn}^{2+}$  ions participate in the formation of the catalytically active complex of the substrate and cotton pyrophosphatase, and the maximum rates of the pyrophosphatase reaction are shown with the  $\text{Mg}^{2+}$  and  $\text{Zn}^{2+}$  ions. In the second place, from the influence of  $\text{Cu}^{2+}$  and  $\text{F}^-$  ions on the enzymatic activity it has been found that they are effective inhibitors that can prevent the activation of cotton pyrophosphatase by bivalent metals.

#### LITERATURE CITED

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